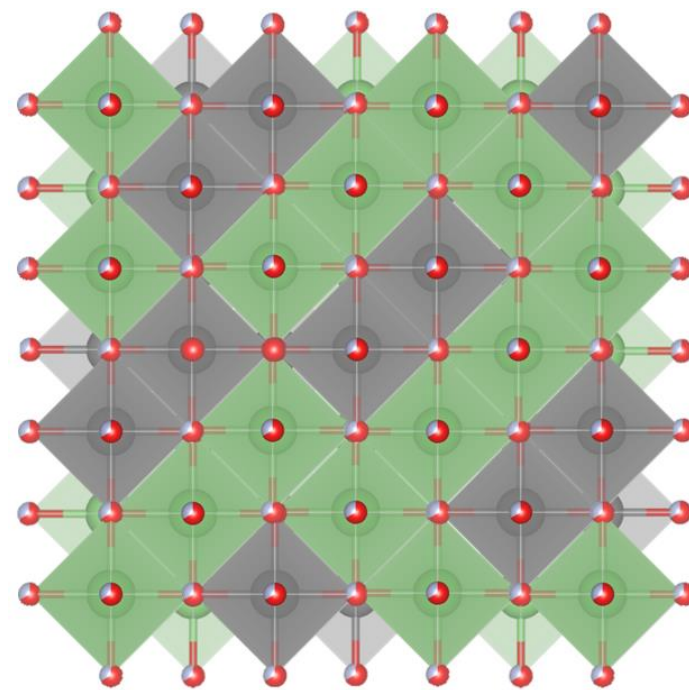
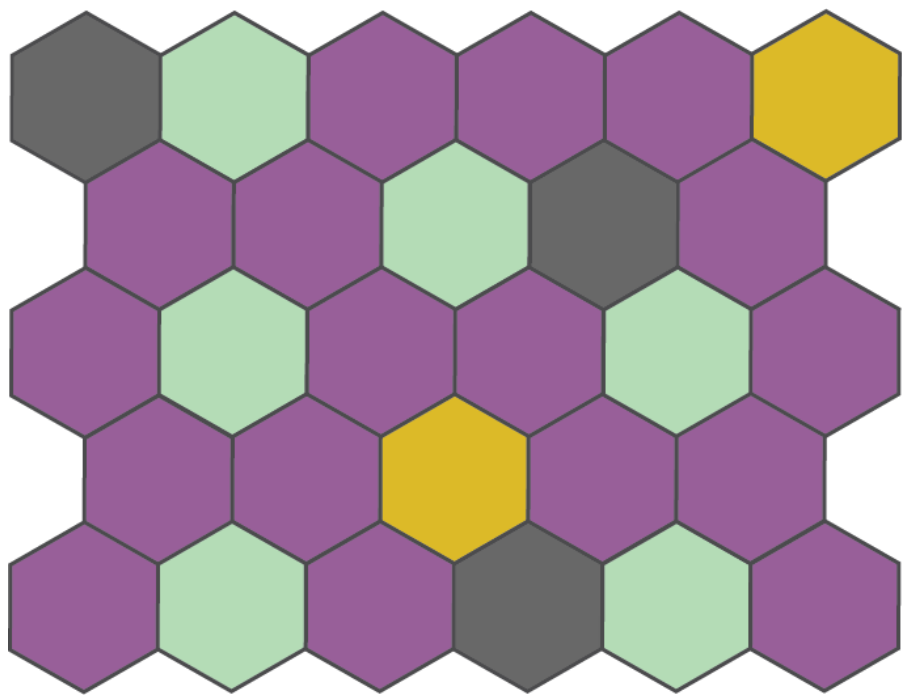




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Non-equilibrium phases and cation mixing in Li-rich rock salt derived positive electrode materials



William Brant
11/05/2021



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Who Am I

Electroactive materials under change

Undergrad University of
Sydney: 2006-2009

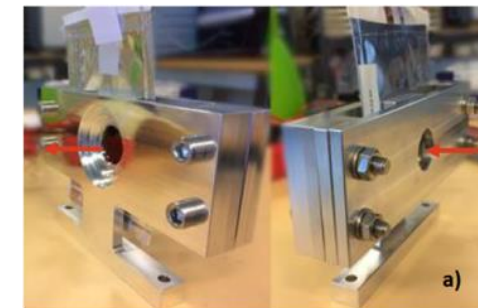
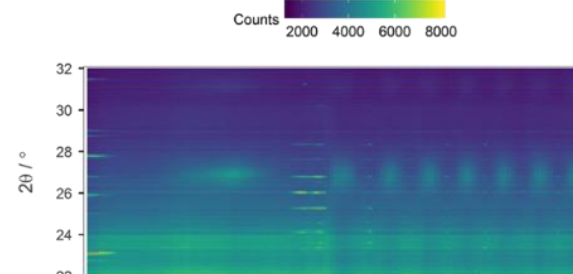
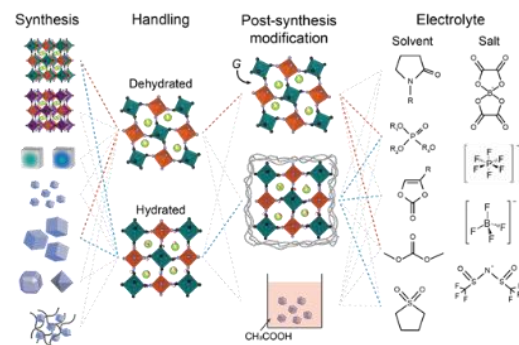
PhD USyd: 2010-2014

Postdoc UU: 2015-2017

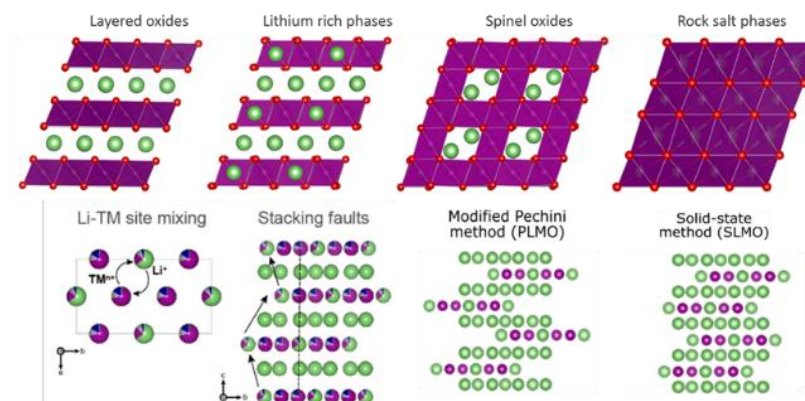
Research leader UU: 2017-now



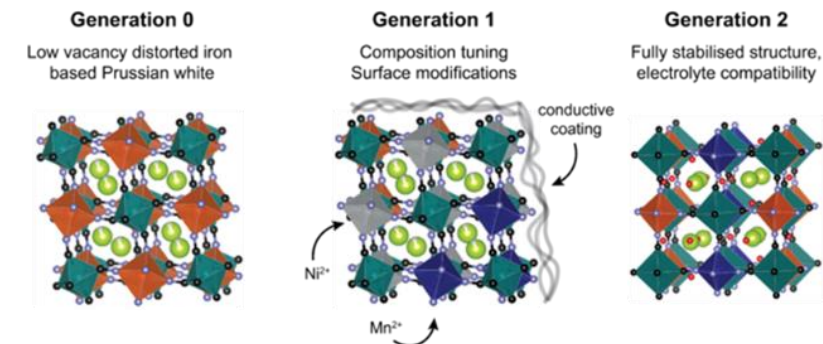
Advanced Synthesis and Characterisation



Rock Salt Oxides



Prussian Blue Analogues





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Who Does the Work

PhDs

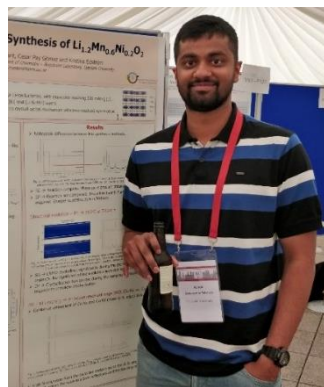
Postdocs

**Undergrad University of
Sydney: 2006-2009**

PhD USyd: 2010-2014

Postdoc UU: 2015-2017

Research leader UU: 2017-now



Ashok Menon



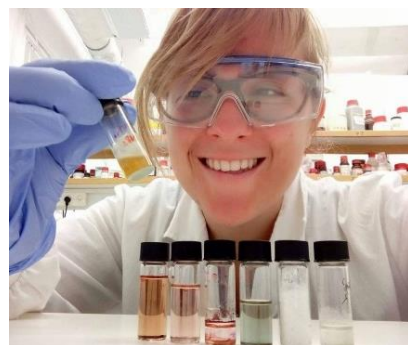
Olle Gustafsson



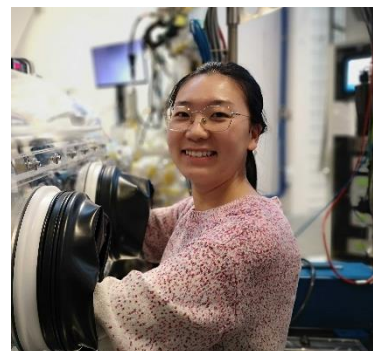
Dickson Ojwang



Adriano Pavan



Djurdjija Dzodan



Heyin Chen

Masters Students

John Corley
Said Khalil

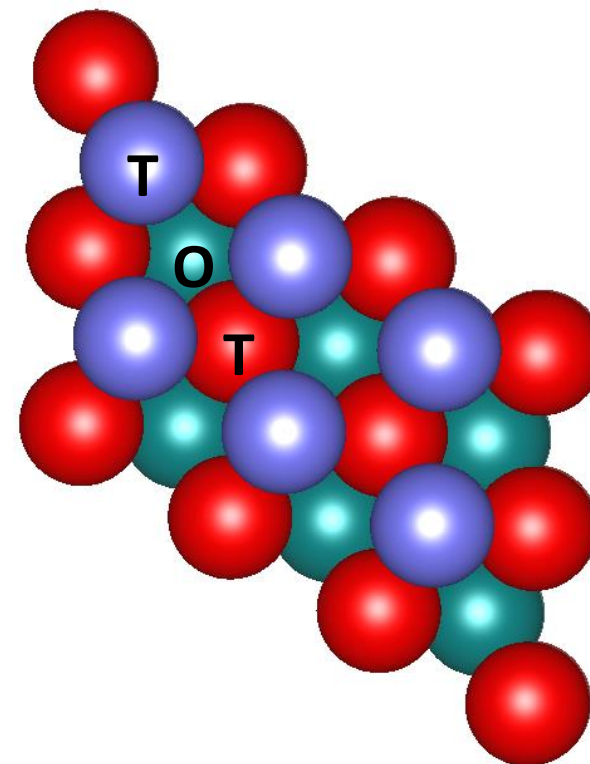
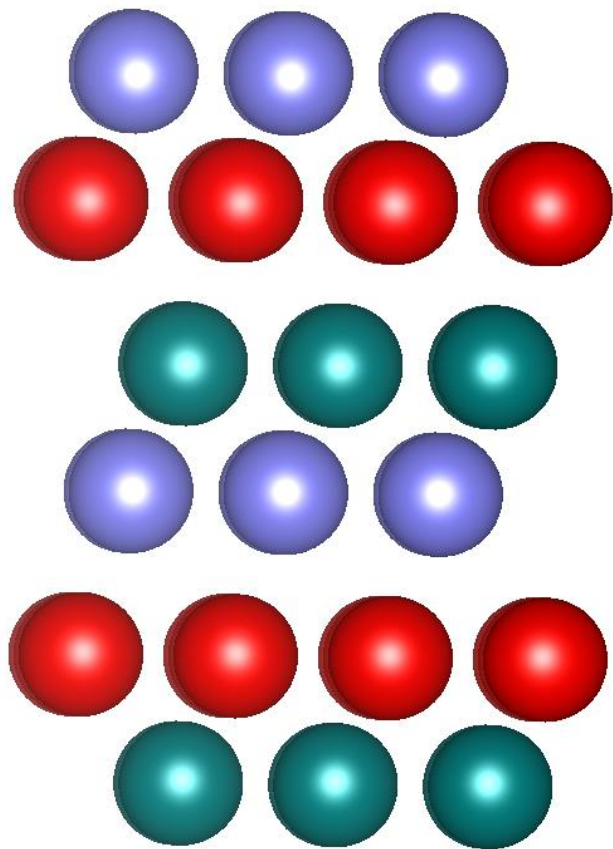
**Materials chemistry and
crystallography**



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Cation order in rock salt phases

AX Cubic close packed array of anions

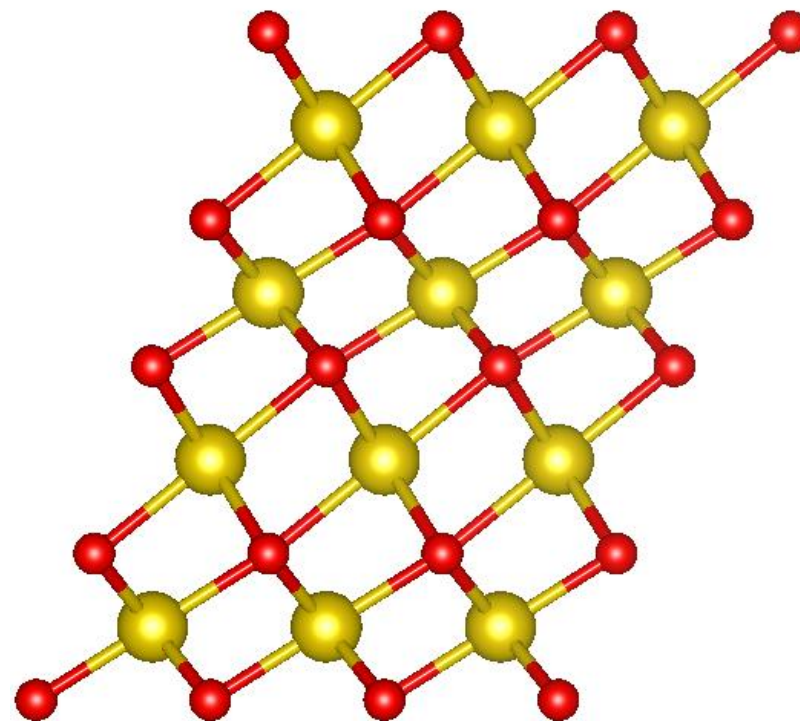
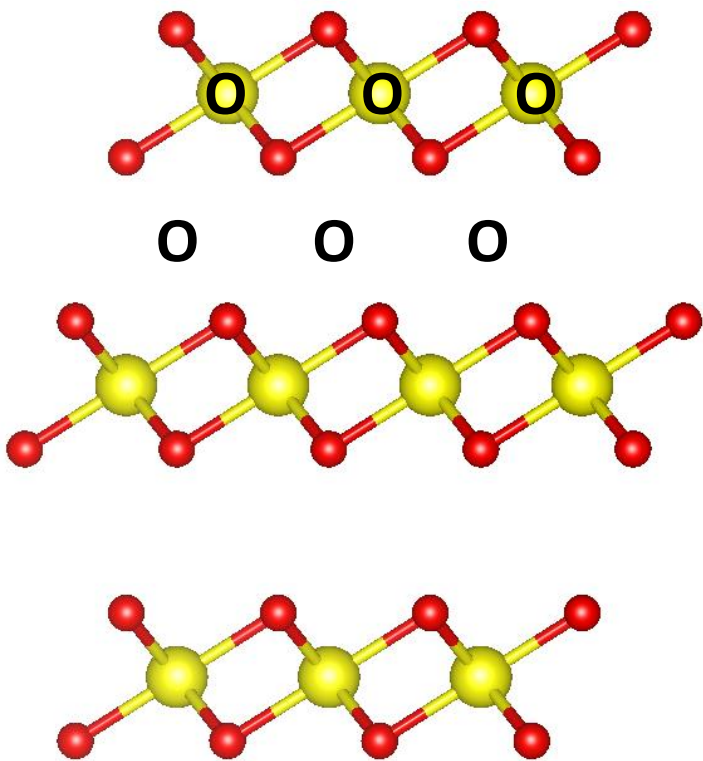




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Cation order in rock salt phases

ABX_2 Cubic close packed array of anions

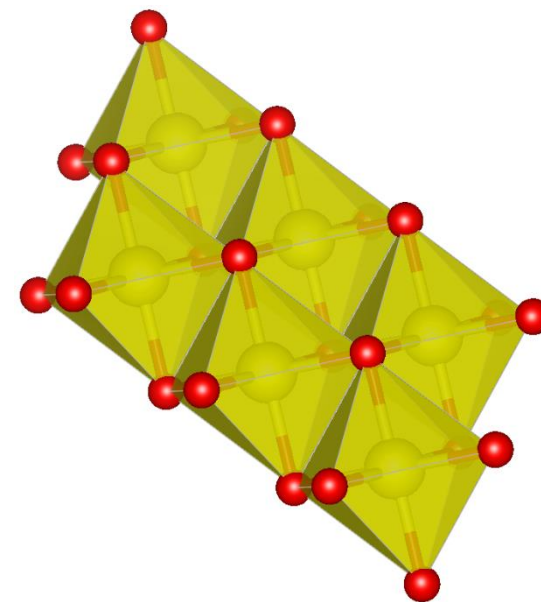
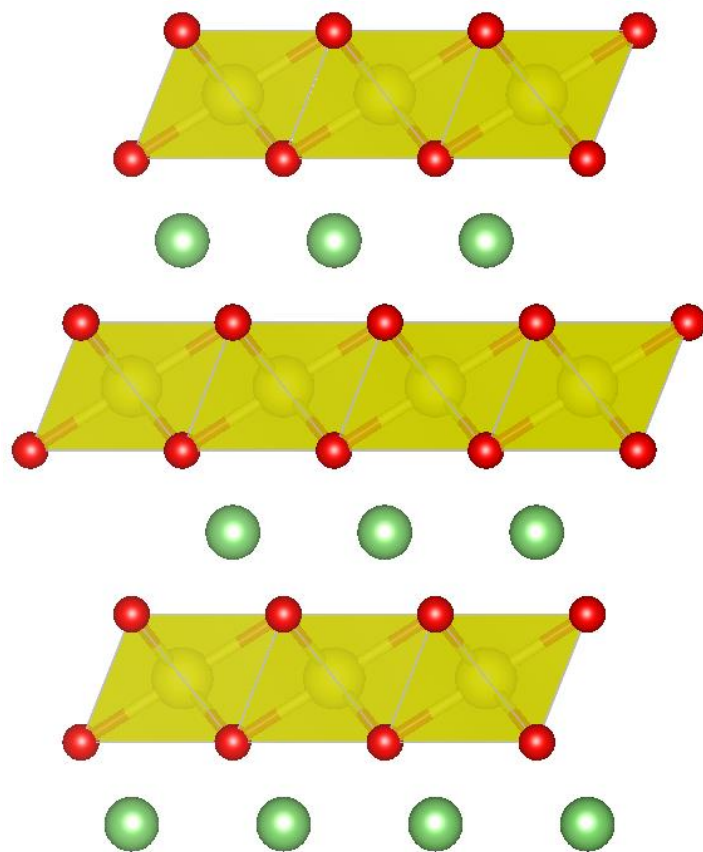




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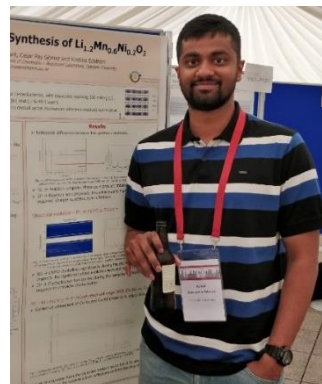
Layered rock salt phases

Li_xMO_2 Cubic close packed array of anions



Li Rich Layered Phases

Compositional complexity and structural analysis



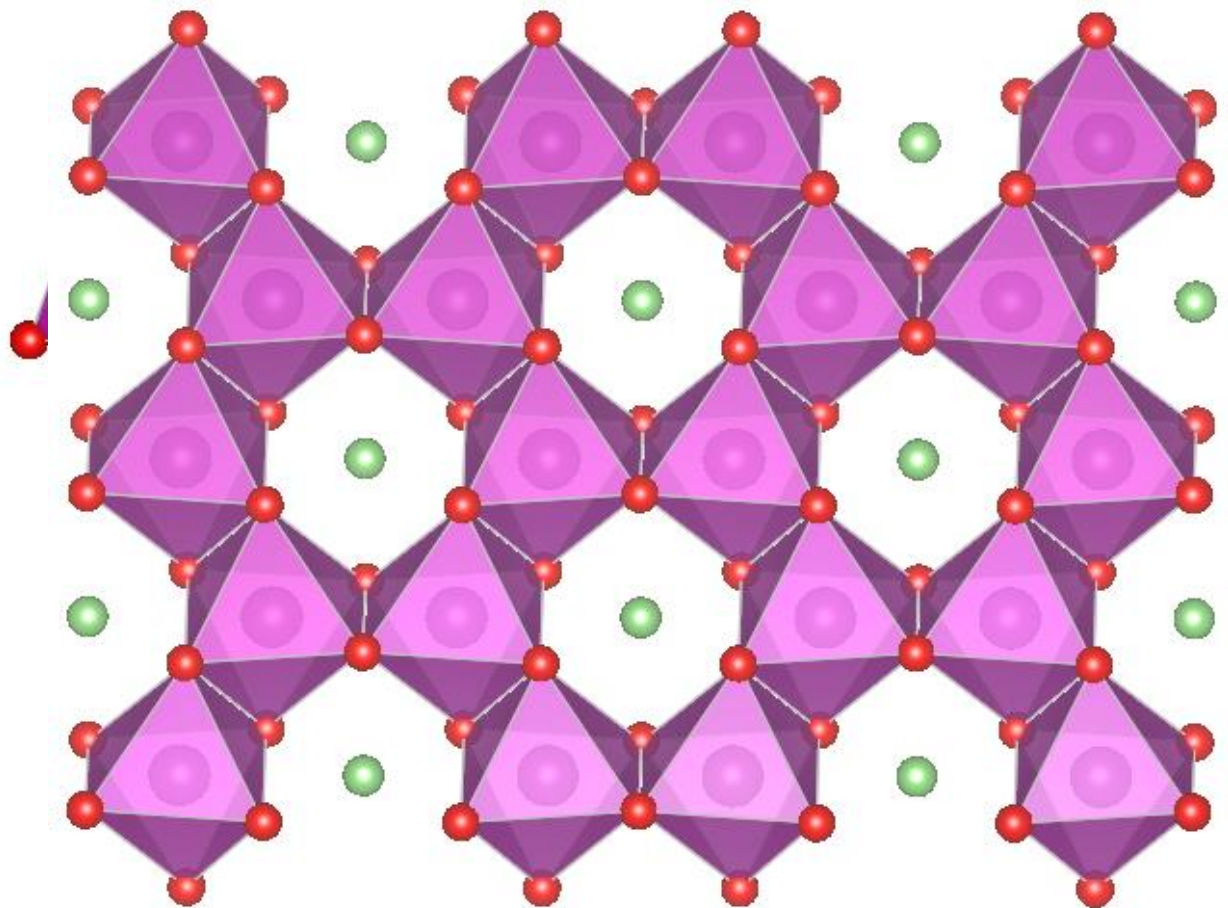


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Li-rich layered rock salt phases



Cubic close packed array of anions



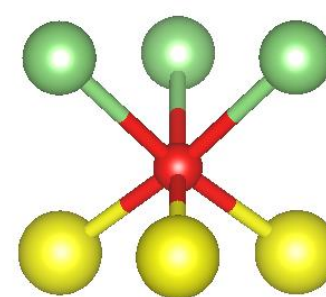
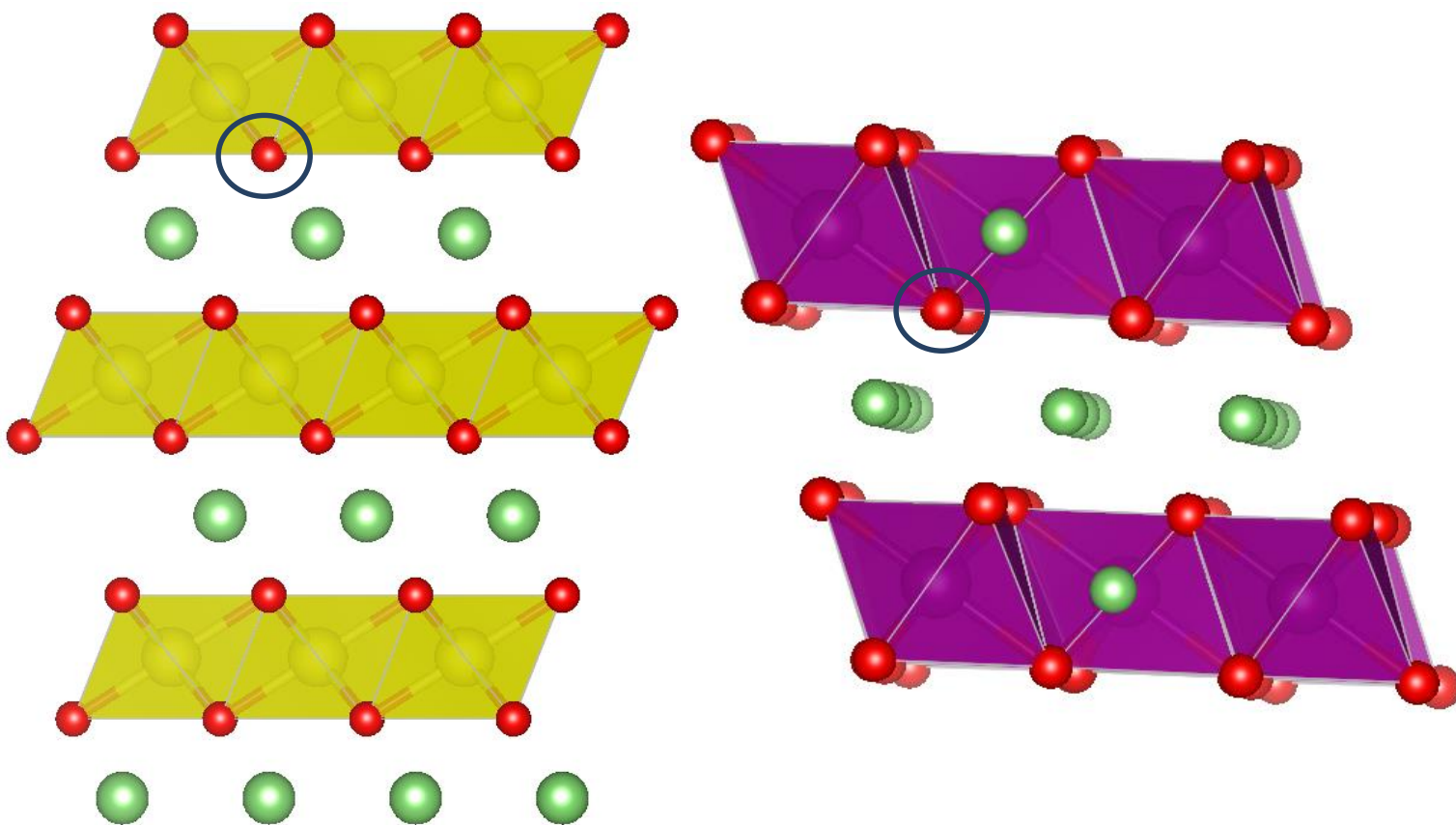


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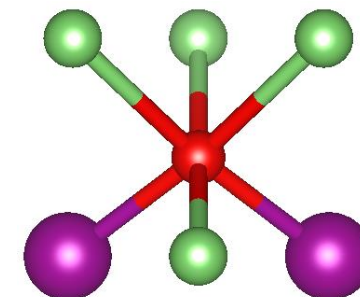
Li-rich layered rock salt phases



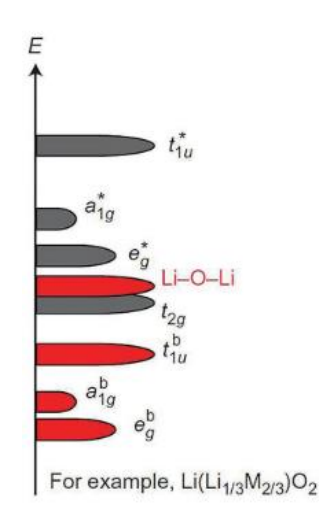
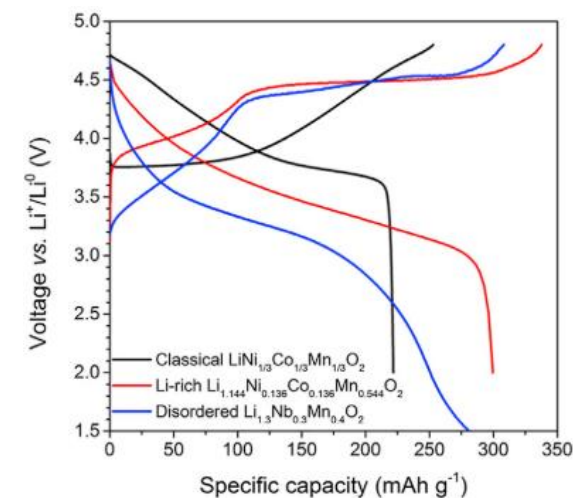
Cubic close packed array of anions



Only M-O-Li



M-O-Li and Li-O-Li





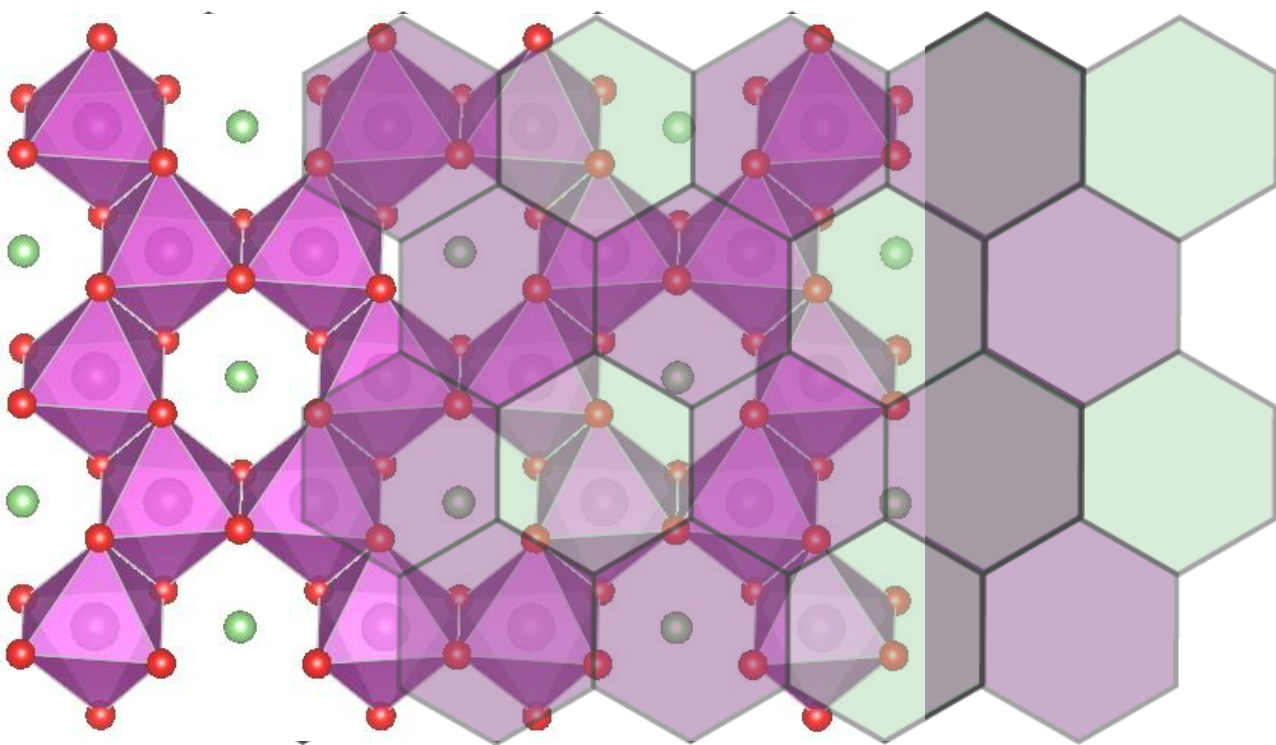
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Structural Complexity in Li_2MnO_3

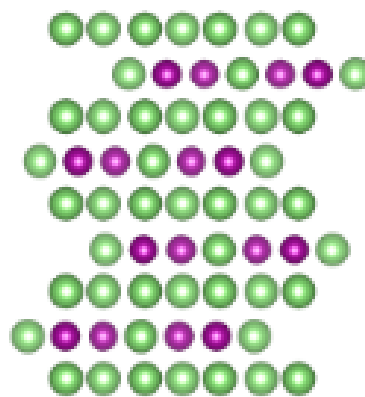
$\text{Li}_{1.33}\text{Mn}_{0.66}\text{O}_2$ – aka Li_2MnO_3

1/3 substitution \rightarrow hexagonal ordering

Mn, Li



Note: Anion sublattice is unchanged



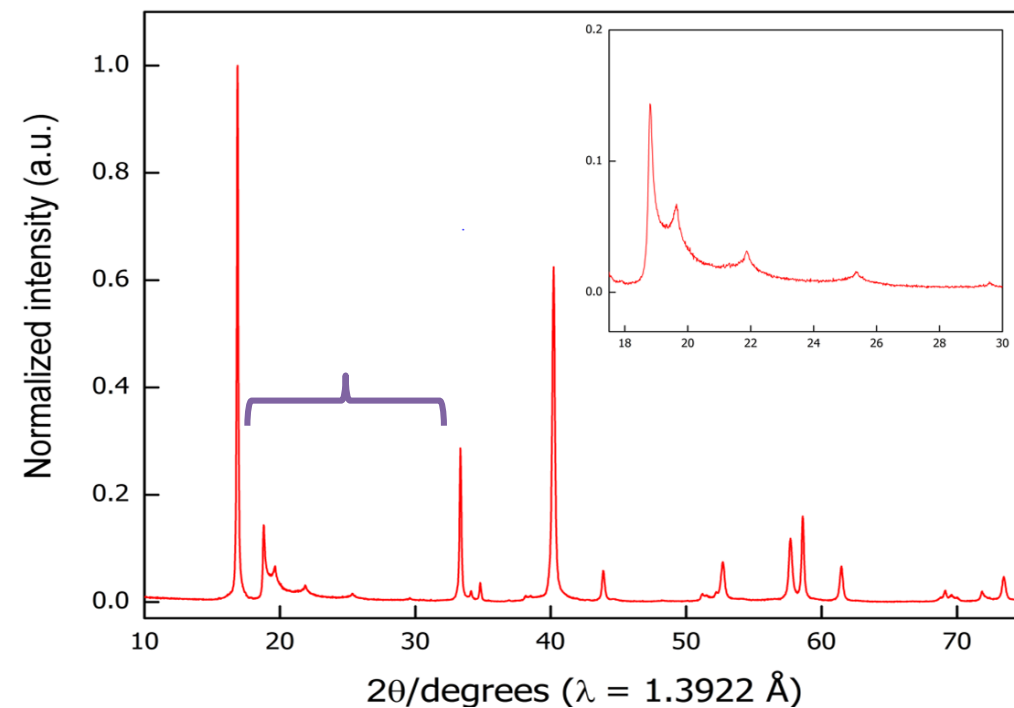
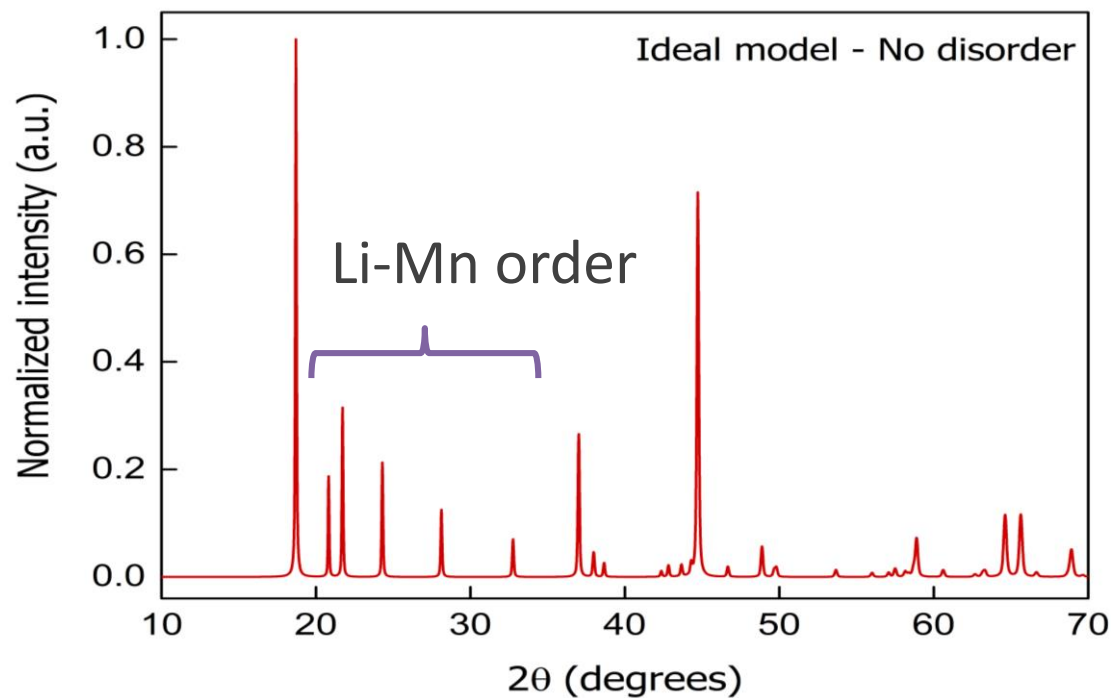
Stacking faults of the
cations only



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Structural Complexity in Li_2MnO_3

Can diffraction see this?

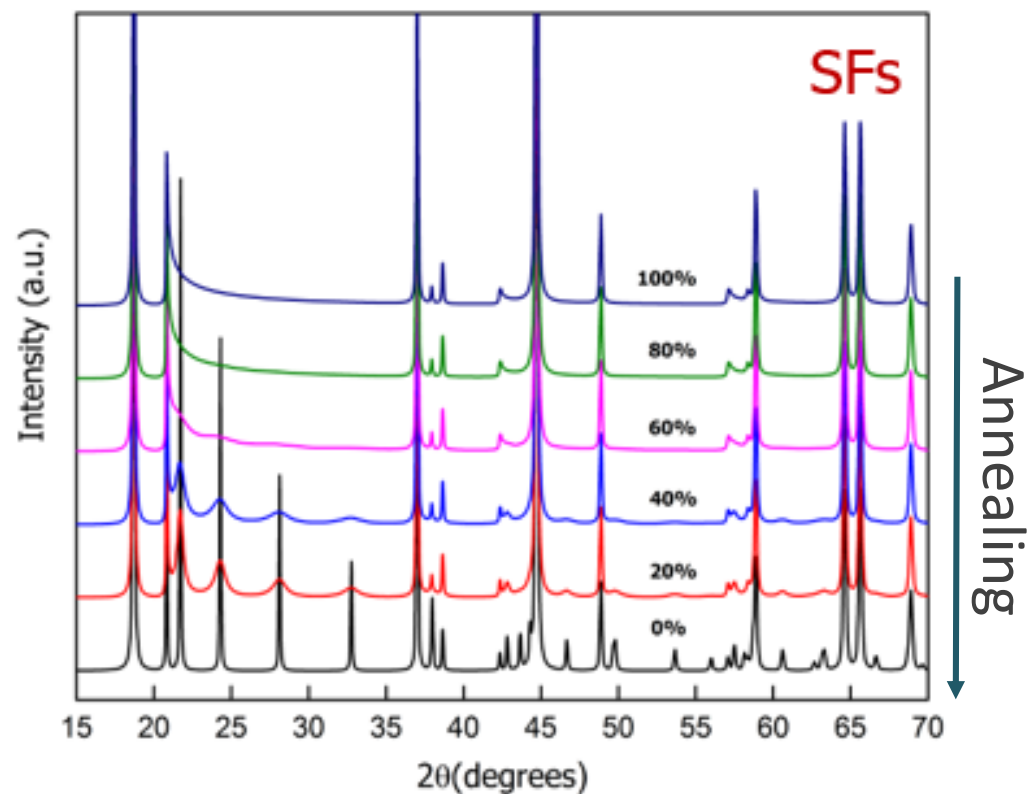




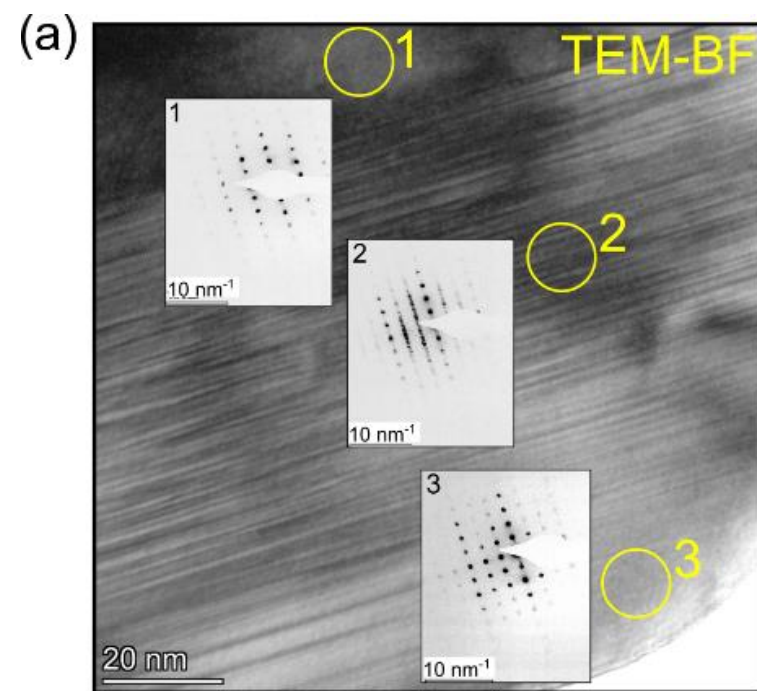
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Structural Complexity in Li_2MnO_3

Remember: The anion sublattice is unchanged!



Consequence: Complete intergrowth of faulted and non-faulted domains

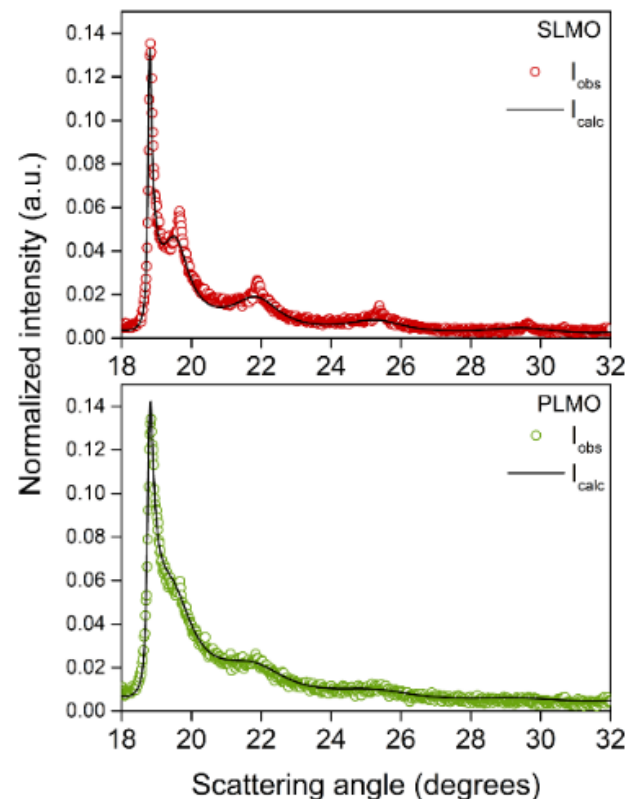
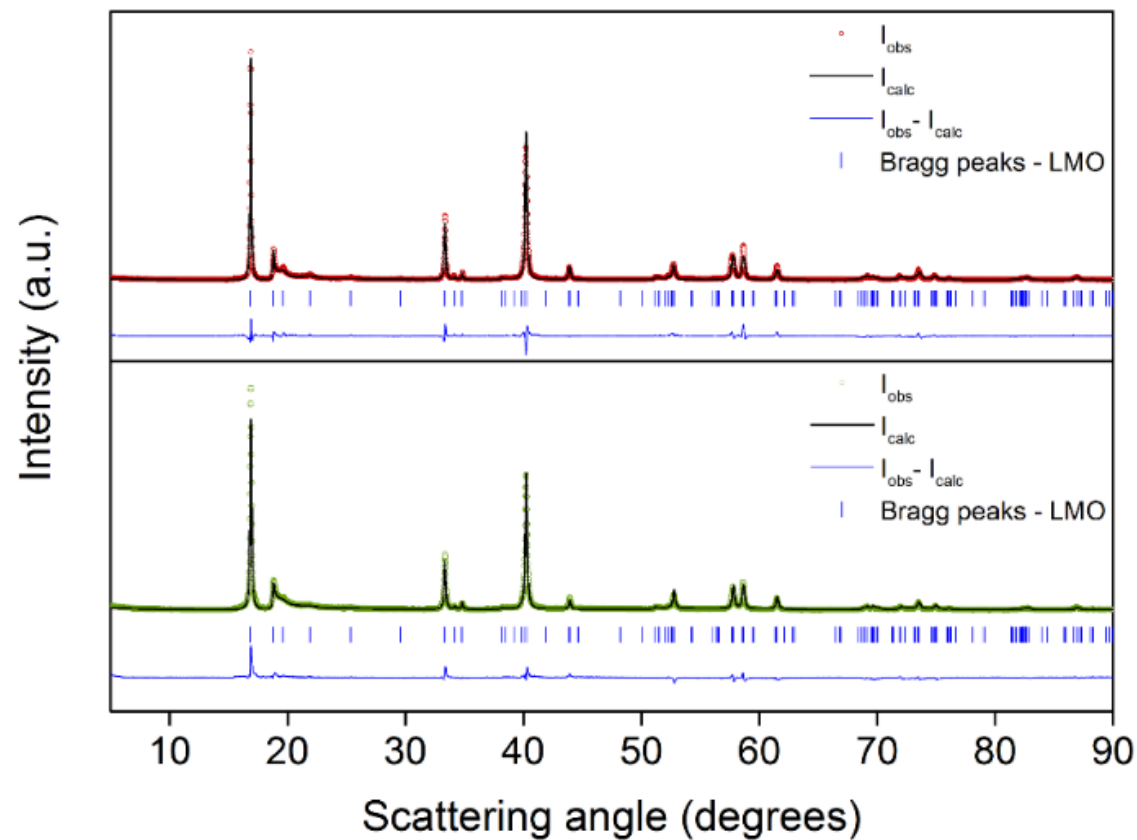




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Structural Complexity in Li_2MnO_3

TEM is localised – is this a long range phenomenon? Can you tell?



Yes! If you actually model the stacking faults

Whether intergrowth occurs depends on synthesis...

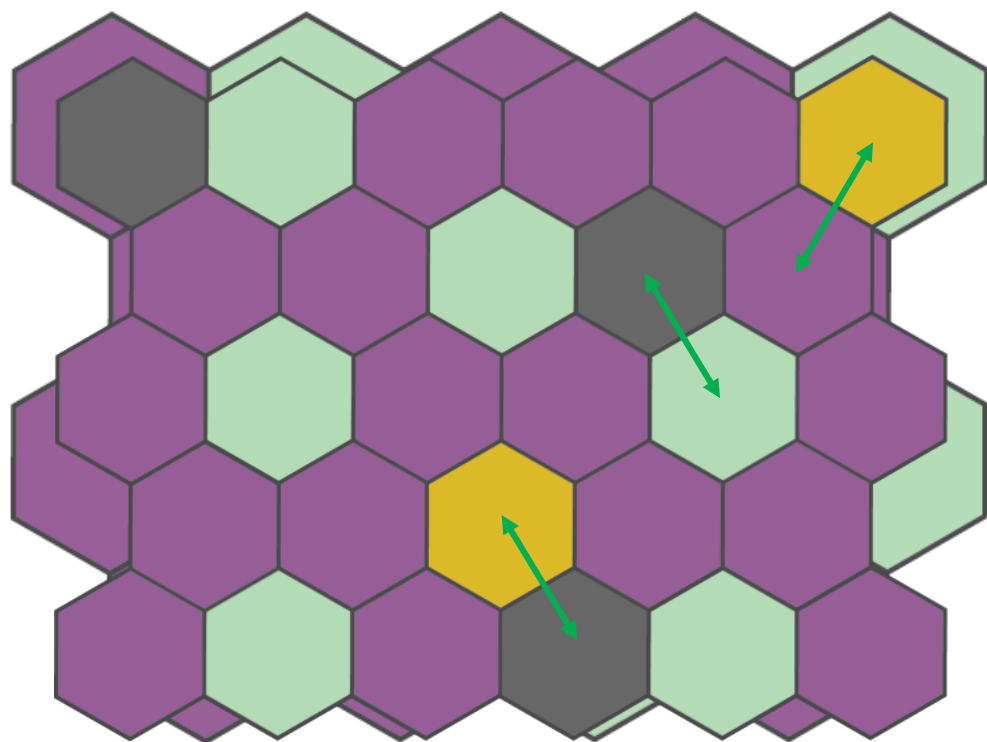


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Why does this matter? – Li_2MnO_3 ($\text{Li}_{1.33}\text{Mn}_{0.66}\text{O}_2$) is compositionally simple, LMNCO is not
→ Three length scales of disorder

Mn, Li, Ni, Co



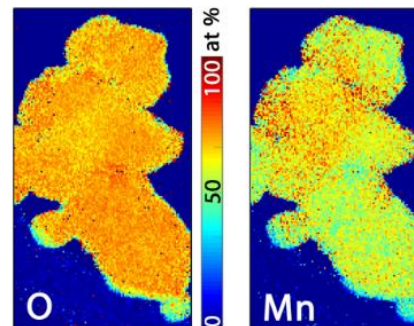
Compositional frustration

Å level – point defects

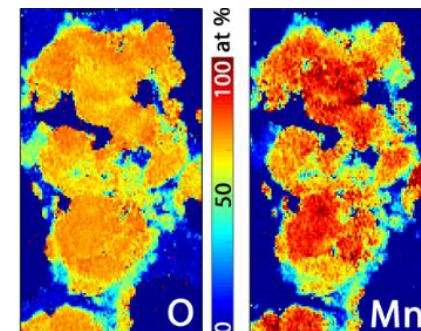
nm level – Stacking faults

μm level – Phase intergrowth

Sol gel



Solid state



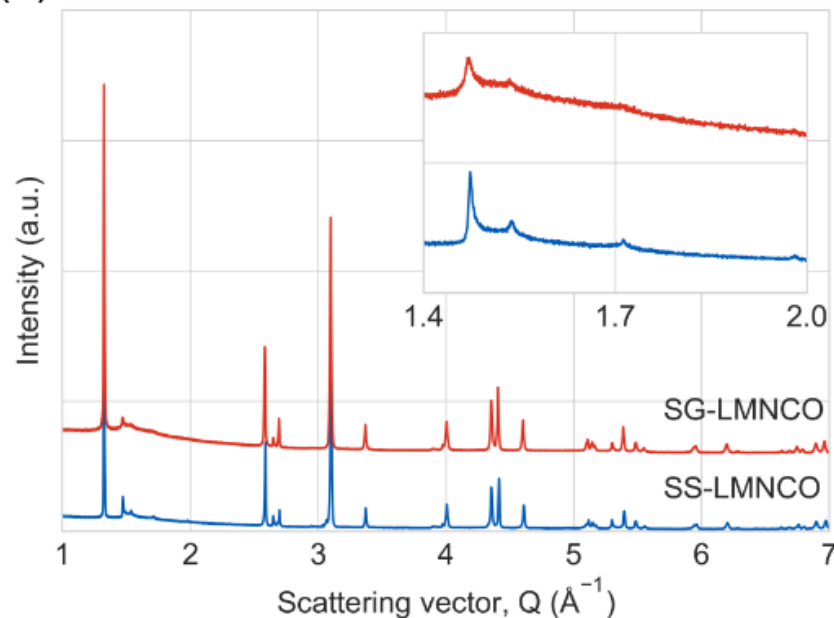


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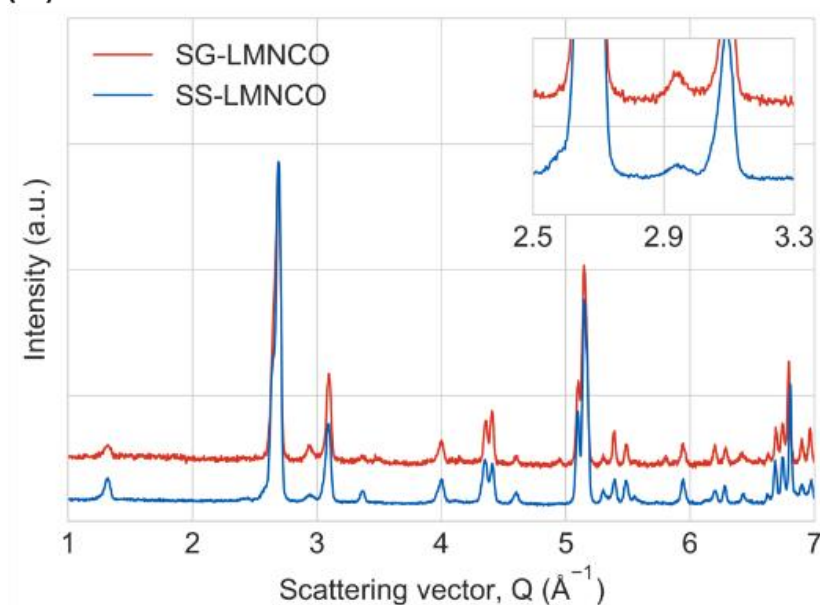


How does the disorder manifest? When regular diffraction breaks down...

(a) XRD



(b) NPD



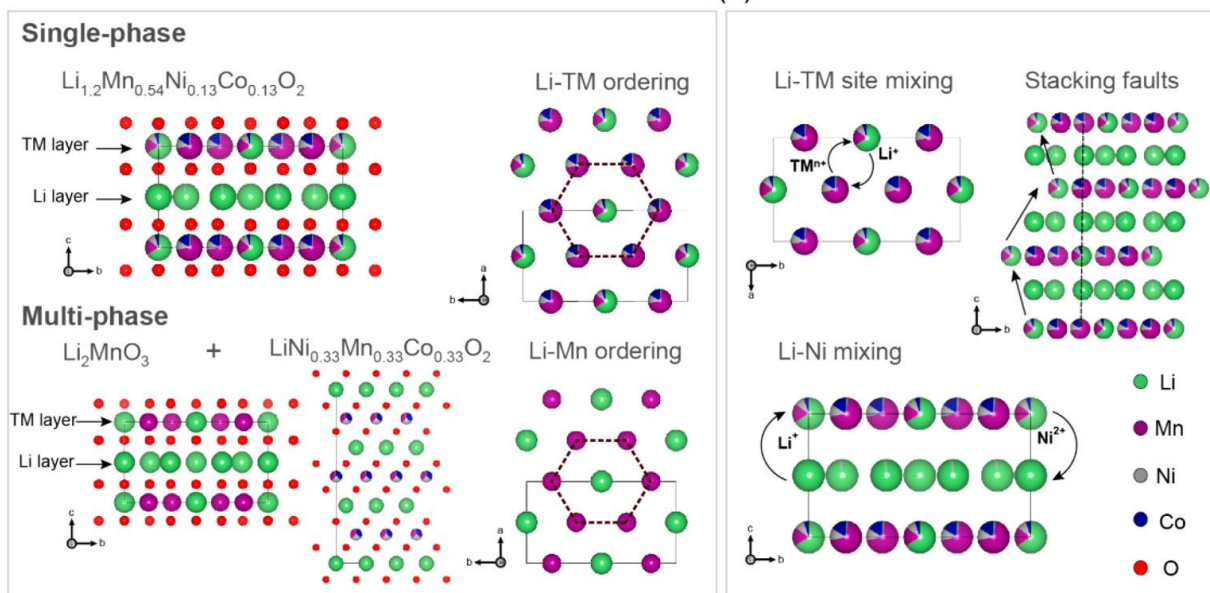


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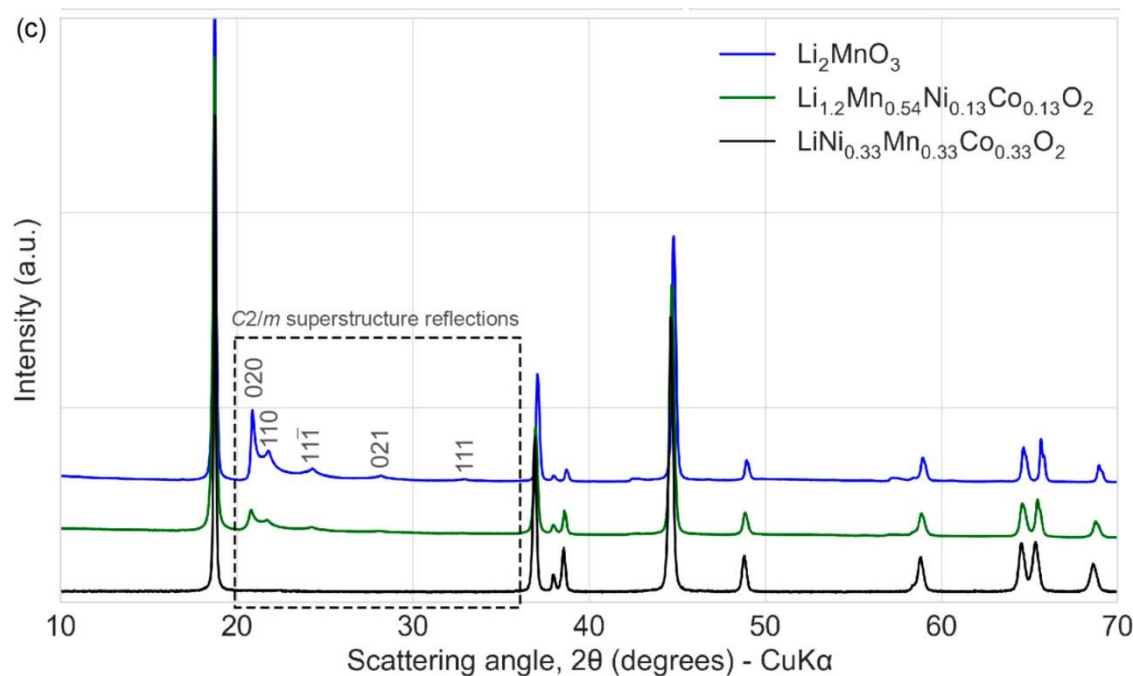


How does the disorder manifest? When regular diffraction breaks down...

(a) LMNCO models



(c)



How can we accurately model these compounds?

X-ray, neutrons, total scattering

Tune into Ashoks Defence on the 14th of June to find out ☺



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Conclusions

The most functionally interesting battery materials are deceptively complex

Properties are dictated by order-disorder transitions

Degree of order and length scale over which (dis)ordering occurs
determined by synthesis

Non-equilibrium phase transitions in **metastable compounds**
makes predicting structural changes in a battery extremely difficult

Neutrons and X-rays are highly complementary and data should be
combined for these systems



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Acknowledgements

Ashok Sreekumar Menon

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Cesar Pay Gomez



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for
ENERGY



**Swedish
Energy Agency**

